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gold(I) dimethylformamide solvate**

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(4,6-Dimercapto-1,3,5-triazine-2-thiolato-*S*²)tris(triphenylphosphine-*P*)-gold(I) dimethylformamide solvate

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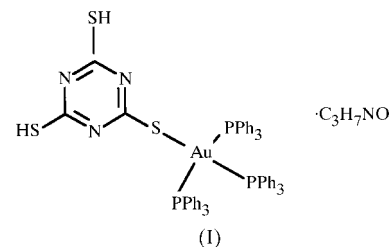
Data validation number: IUC0000187

In the title compound, $[\text{Au}(\text{C}_3\text{H}_2\text{N}_3\text{S}_3)(\text{C}_{18}\text{H}_{15}\text{P}_3)] \cdot \text{C}_3\text{H}_7\text{NO}$, the Au^{I} atom has a distorted tetrahedral geometry consisting of one uncoordinated dimethylformamide molecule, one trithiocyanurate ligand and three PPh_3 ligands. The $\text{S}-\text{Au}$ distance is 2.909 (2) Å, and the $\text{P}-\text{Au}$ distances fall in the range 2.400 (14)–2.4074 (13) Å. The $\text{S}-\text{Au}-\text{P}$ angles are 92.81 (6), 99.17 (6) and 104.21 (5)°, and the $\text{P}-\text{Au}-\text{P}$ angles are 114.46 (5), 119.04 (5) and 119.66 (5)°.

Comment

There has been growing interest in the study of Cu^{I} , Ag^{I} and Au^{I} complexes, and in particular polynuclear systems; this stems from the tendency of these metal ions to form clusters and aggregates as a result of weak metal–metal interactions (Mingos, 1976; Schmidbaur, 1995) and the recent demonstration that a number of these aggregates exhibit rich luminescence behaviour (Vickery *et al.*, 1997; Yam *et al.*, 1999). Until now, Au^{I} -sulfido complexes have been comparatively rare (Canales *et al.*, 1994). Recently, an interesting example was found in studies of trigold(I) complexes of trithiocyanuric acid, in which an unexpected reaction led to formation of an interesting sheet structure through multiple intermolecular $\text{Au} \cdots \text{Au}$ -bonded self-assembly (William *et al.*, 1999). Also by use of Au^{I} , PPh_3 and trithiocyanuric acid, we obtained a different product, namely (4,6-dimercapto-1,3,5-triazine-2-thiolato-*S*²)tris(triphenylphosphine-*P*)gold(I) dimethylformamide (DMF) solvate, (I), including a DMF molecule, one Au^{I} atom, one trithiocyanurate ligand and three PPh_3 ligands. In the crystal structure, only one thiolate of the trithiocyanuric acid has been coordinated to the Au^{I} atom and each Au^{I} is coordinated by one thiolate group of the trithiocyanurate ligand and three P atoms of the PPh_3 groups, giving a tetrahedral coordination. The $\text{S}-\text{Au}$ distance is 2.909 (2) Å, which is much longer than the normal $\text{S}-\text{Au}$ distance, and the $\text{P}-\text{Au}$ distances fall in the range 2.400 (14)–2.4074 (13) Å. The $\text{S}-\text{Au}-\text{P}$ angles are 92.81 (6), 99.17 (6) and 104.21 (5)°, and

the $\text{P}-\text{Au}-\text{P}$ angles are 114.46 (5), 119.04 (5) and 119.66 (5)°. Thus, the geometry is distorted tetrahedral.

**Experimental**

The title compound, (I), was obtained from the reaction of trithiocyanuric acid, PPh_3 and AuCN in DMF. To a solution of trithiocyanuric acid (0.5 mmol, 0.09 g) in DMF, AuCN (20 ml, 0.5 mmol, 0.12 g) and PPh_3 (1 mmol, 0.26 g) were added. After stirring for 2 h, the orange solution was filtered. Ether was diffused slowly into the filtrate solution. A week later, crystals suitable for X-ray diffraction analysis were obtained.

Crystal data

$[\text{Au}(\text{C}_3\text{H}_2\text{N}_3\text{S}_3)(\text{C}_{18}\text{H}_{15}\text{P}_3)] \cdot \text{C}_3\text{H}_7\text{NO}$
 $M_r = 1233.13$
 Triclinic, $P\bar{1}$
 $a = 13.7053$ (12) Å
 $b = 13.9751$ (12) Å
 $c = 14.4614$ (12) Å
 $\alpha = 80.665$ (2)°
 $\beta = 87.283$ (2)°
 $\gamma = 87.942$ (2)°
 $V = 2729.0$ (4) Å³

$Z = 2$
 $D_x = 1.501$ Mg m⁻³
 Mo $K\alpha$ radiation
 Cell parameters from 925 reflections
 $\theta = 11.6$ – 18.9 °
 $\mu = 2.943$ mm⁻¹
 $T = 293$ (2) K
 Block, colourless
 $0.25 \times 0.22 \times 0.20$ mm

Data collection

SMART CCD diffractometer
 ω scans
 Absorption correction: multi-scan (*XEMP* in *SHELXTL*; Sheldrick, 1997)
 $T_{\text{min}} = 0.420$, $T_{\text{max}} = 0.555$
 18 651 measured reflections

12 409 independent reflections
 9513 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.0353$
 $\theta_{\text{max}} = 27.56$ °
 $h = -17 \rightarrow 17$
 $k = -18 \rightarrow 12$
 $l = -18 \rightarrow 18$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.045$
 $wR(F^2) = 0.113$
 $S = 0.967$
 12 409 reflections
 624 parameters

H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0602P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} = 0.002$
 $\Delta\rho_{\text{max}} = 1.19$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.81$ e Å⁻³

Data collection: *SMART* (Siemens, 1994); cell refinement: *SMART*; data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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References

- Canales, F., Gimeno, M. C. & Jones, A. L. (1994). *Angew. Chem. Int. Ed. Engl.* **33**, 769–770.
- Mingos, D. M. P. (1976). *J. Chem. Soc. Dalton Trans.* pp. 1163–1169.
- Schmidbaur, H. (1995). *Chem. Soc. Rev.* pp. 391–400.
- Sheldrick, G. M. (1997). *SHELXTL*. Version 5.10. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1994). *SMART* and *SAINT*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Vickery, J. C., Olmstead, M. M. & Fung, A. L. B. (1997). *Angew. Chem. Int. Ed. Engl.* **36**, 1179–1181.
- William, J. H., Michael, C. J. & Richard, J. P. (1999). *Inorg. Chem.* **38**, 5930–5931.
- Yam, V. W.-W., Cheng, E. C.-C. & Cheung, K.-K. (1999). *Angew. Chem. Int. Ed. Engl.* **38**, 197–199.